Environment Effects on the Optical Properties of a Rhodamine Heterodimer-DNA Supramolecular Complex: A computational and Experimental Study

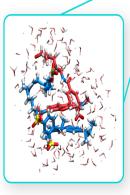
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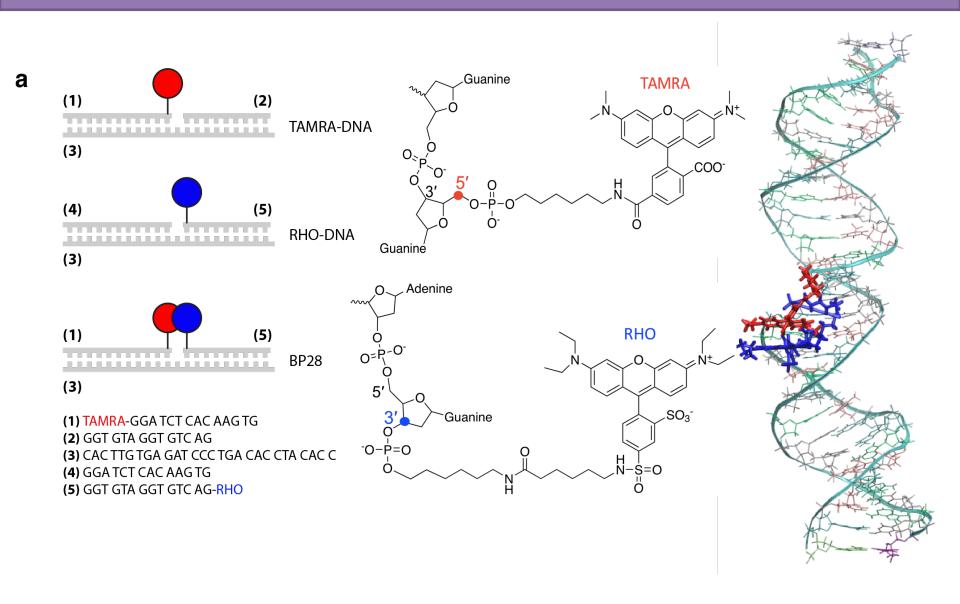
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RHO–TAMRA STRUCTURE



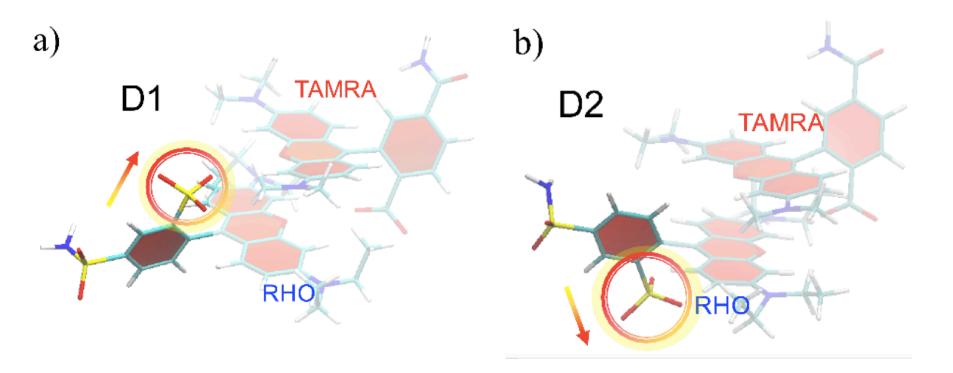
Monomers RHO (red) TAMRA (blue) 1.0 Absorption cross-section, a.u. 1.0 Absorption cross-section, a.u. 0.8 0.8 0.6 0.6 0.4 0.4 0.2 0.2 0.0 1.7 1.6 1.8 1.9 2.0 2.1 2.2 1.7 1.8 1.9 2.0 2.1 1.6 x10⁴ ω, cm⁻¹ ω**, cm**-1 x10⁴

TDDFT/ CAMD-B3LYP/6-311(d,p)/PCM, harmonic oscillator model (M. Lax 1952)

Heterodimer RHO–TAMRA

- Reproduce correctly the spectrum lineshape including the complex media
- Develop the approach for multi-mode harmonic oscillator model including complex media
- Describe correctly active normal modes
- Define the sampling of MD trajectory for the nuclearensemble approximation

Two types of conformers

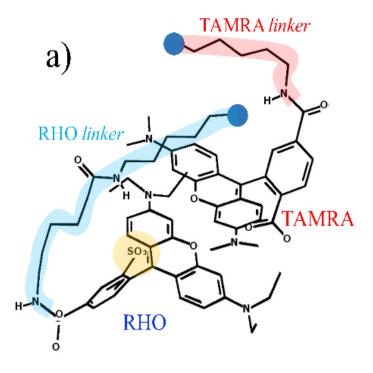


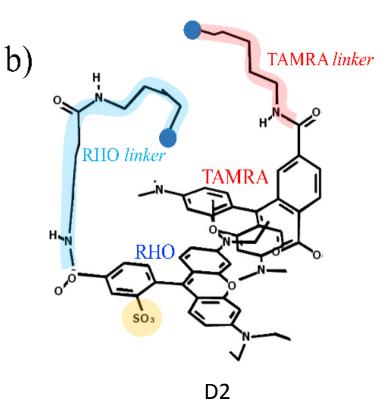
The main difference in the orientation of the anionic sulfo-group of the RHO unit inside (a) and outside (b) the dimer subspace.

We denote these structures as isomers because they do not interconvert within 60ns MD trajectory at room temperature.

- PARM99/BSC0 force field for DNA,
- GAFF force field for the chromophores and the aliphatic linkers
- Cubic periodic box with waters (TIP3P) with a minimum buffer 10 Å between the atoms of DNA or chromophore and the edge of the box. Sodium counterions were added (27 atoms) to ensure the neutrality of the system

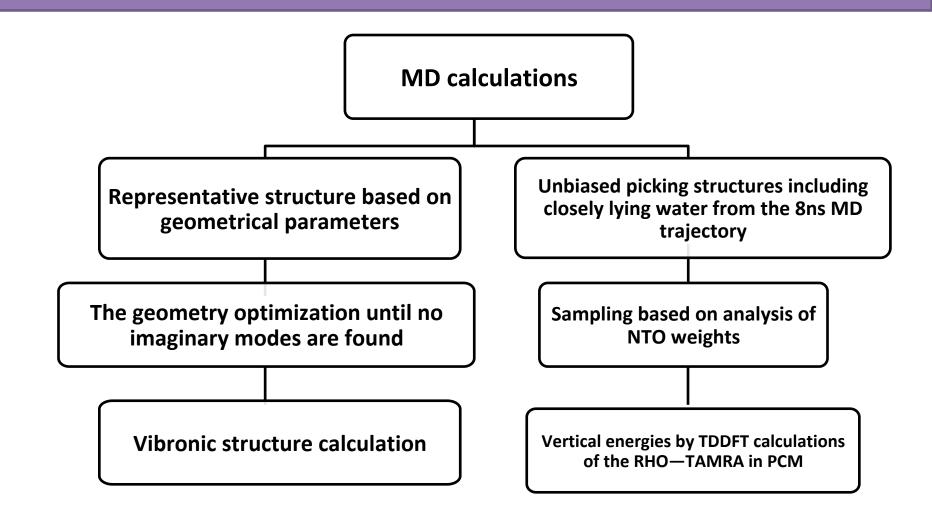
- The duration of the production trajectory: 8 ns (time step 2 fs)
- This step was performed in the canonical NVT ensemble
- Short-range interactions were truncated at a cutoff distance of 9 Å.



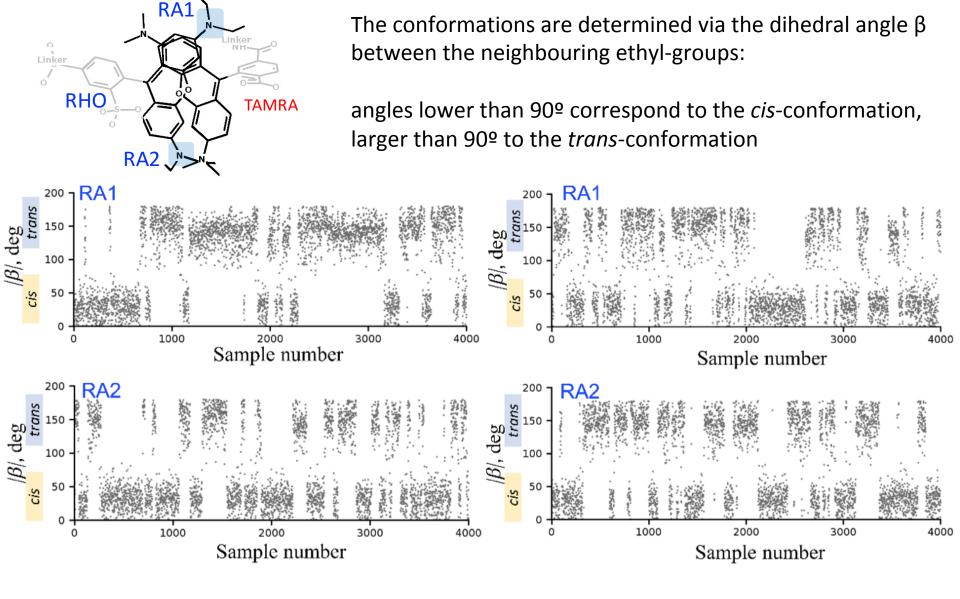


D1

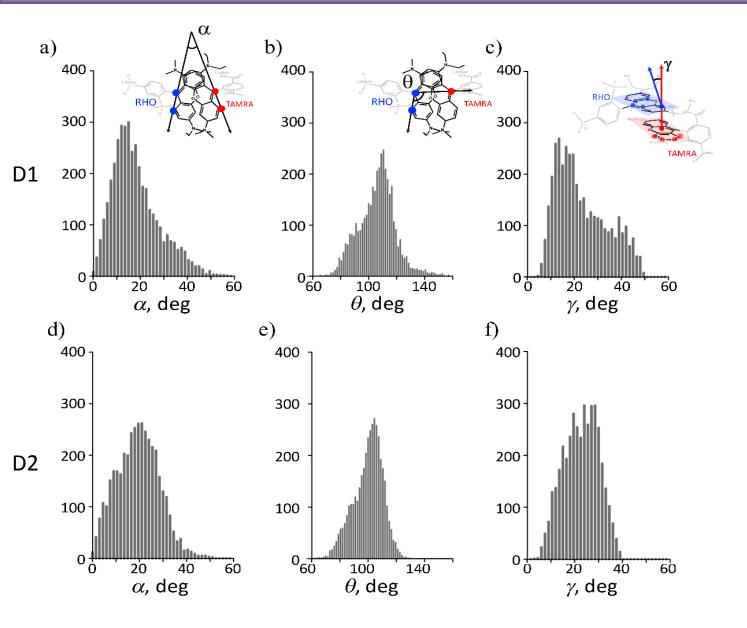
COMPUTATIONAL DETAILS



Rhodamine amino-groups conformations along the MD trajectory 9



Analysis of intermolecular angles along the MD trajectory

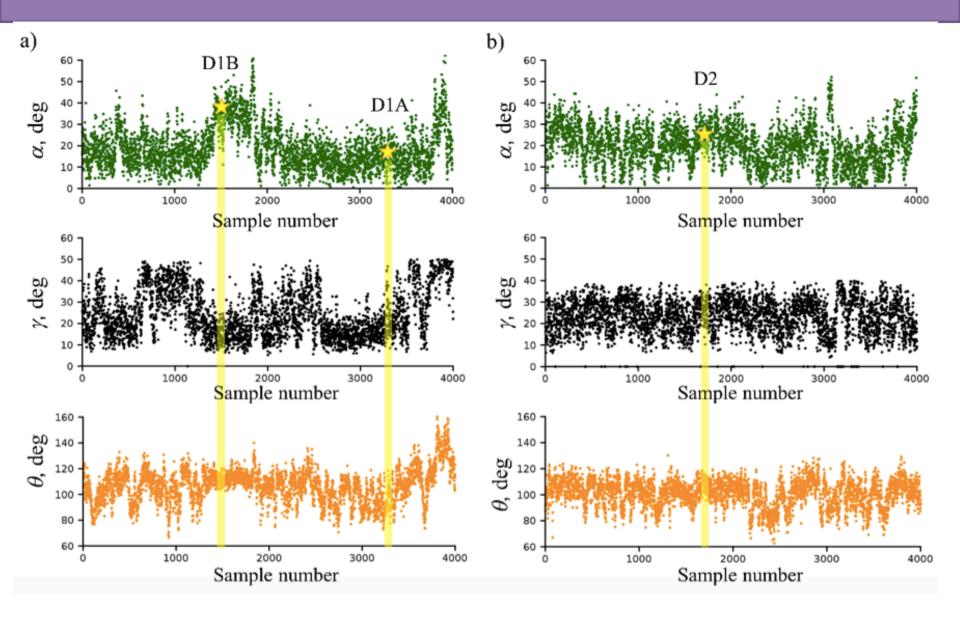


α- angle between
 the polarization
 axes of the two
 units

θ- angle made by
the polarization
axes of the unit
molecule with the
line of molecular
centre

Y- mean angle between the planes of the xanthene moieties

Analysis of intermolecular angles along the MD trajectory



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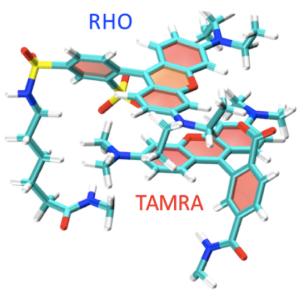
Scheme of QM/MM (ONIOM 2-layers model)

QM + MM Frozen water sphere ~27 Å

MM: PARM99/BSC0 force field for DNA, TIP3P - for waters, GAFF - for the chromophores and linkers

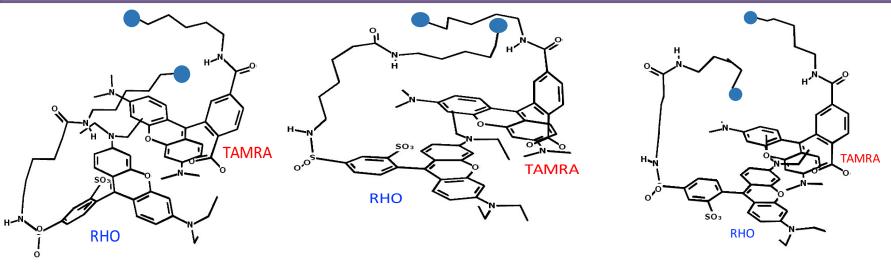
Unfrozen part of QM + MM Water sphere ~7 Å

QM: ground state equilibrium geometry and normal modes: DFT-D/CAM-B3LYP/ 6-311G (d,p) with electrostatic embedding *excitation energies, dipoles, oscillator strengths and gradients* in QM/MM: TDDFT/CAM-B3LYP/6-311G (d,p)



QM part

Conformers along the MD trajectory



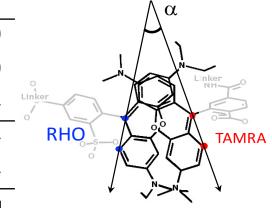
D1A (α = 21°)

D1B ($\alpha = 35^{\circ}$)

D2 ($\alpha = 25^{\circ}$)

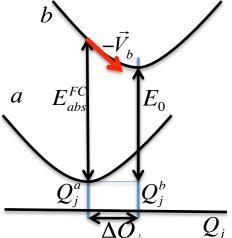
	α,	δE(GS),	ΔE(S ₁ -S ₂),	S_0 - S_1		S ₀ -S ₂		_
	degrees	kcal/mol	cm⁻¹	E _{exc} , eV	f	E _{exc} , eV	f	
D1A/QM/MM	21	0	1290	2.55	0.76	2.71	1.80	-
D1B/QM/MM	35	8	2016	2.47	0.95	2.72	1.50	0
D2/QM/MM	27	13	1210	2.57	0.70	2.72	1.84	Linke
D1/PCM	3	0*	1613	2.58	0.57	2.78	1.94	
D2/PCM	5	5*	1613	2.58	0.47	2.78	2.04	_

^{*}Ground state energies of the bare dimers within PCM solvation are computed relative to the most stable bare dimer, D1/PCM.



for normal modes

Harmonic approximation $|\langle \chi_{a0} | \chi_{bn} \rangle|^2$ - Franck-Condon factors



Large number of vibrational states – need to use some approximations:

- **Condon** approximation
- Parallel harmonic approximation
- No Duschinsky rotation effect
- Temperature effect: Boltzmann distribution in

the ground state

Lax M. J. Chem. Phys. 1952. V. 20. Pp. 1752. Dispersion of the absorption band

$$\Delta E_{ab}$$

 $\left(\delta v\right)^2 = \sum \left(\delta v_j\right)^2 = \sum S_j \Omega_j^2 \operatorname{cth}\left(\frac{\hbar \Omega_j}{2kT}\right)$ $\Gamma = 2\sqrt{2\ln 2} \cdot \left(\delta v\right)^{2j}$ Contribution of each normal mode to the broadening: $(\delta v_j)^2 / (\delta v)^2$ $S_j = \frac{M\Omega_j (\Delta Q)^2}{2\hbar}$ - Huang-Rhys factors

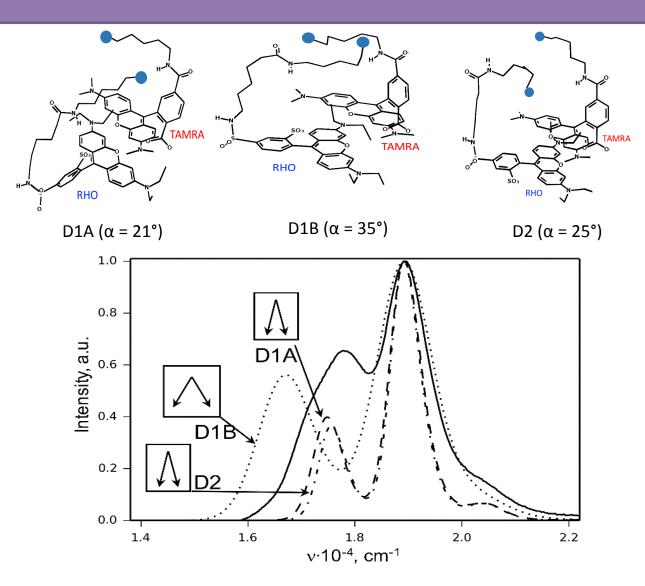
Kretov MK, et al. (2012) J Lumin, 132(8):2143.

Input

Normal modes of the ground state **Electronic excitation energies Electronic transition dipoles** Gradients on the excited state potential

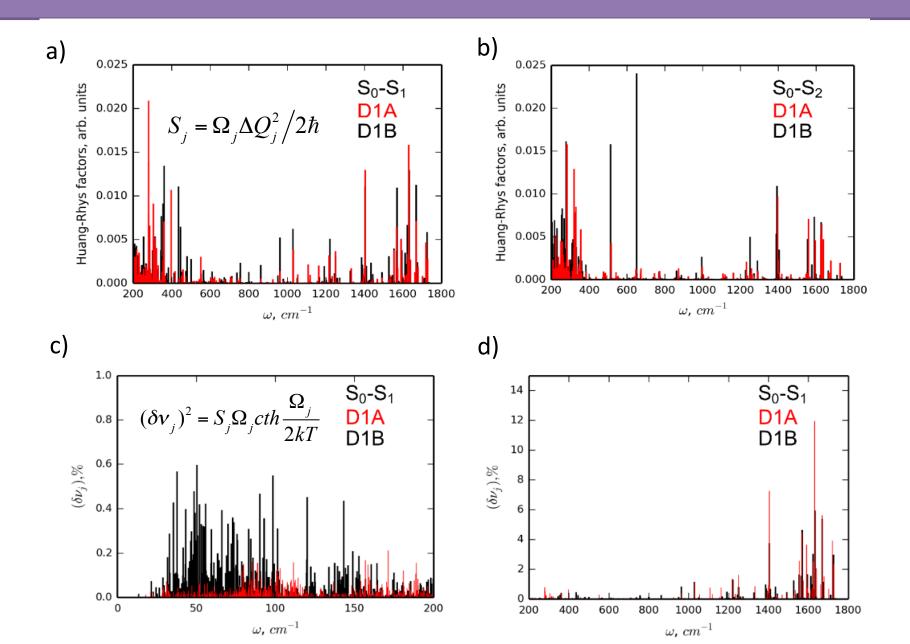
> Optically active vibrational modes can be defined

Absorption lineshape of D1 and D2 conformers

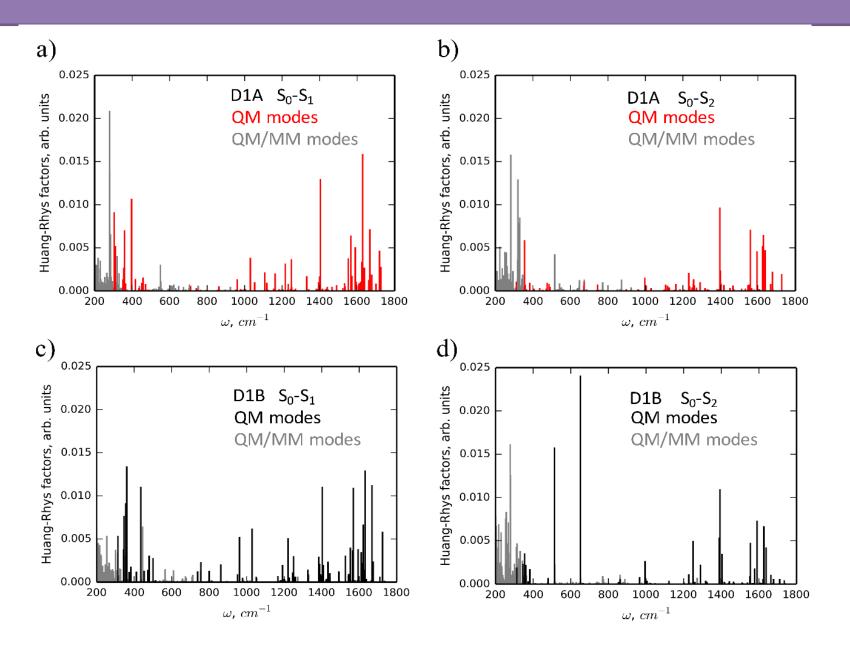


 Δ (D1A) = -3086 cm⁻¹, Δ (D1B) = -3070 cm⁻¹ and Δ (D2) = -2765 cm⁻¹

Analysis of the optically active normal modes

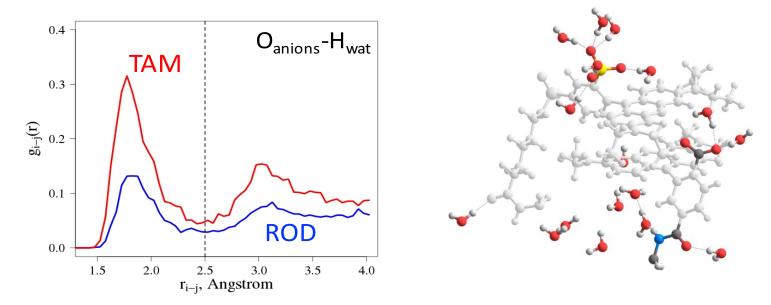


Analysis of the optically active normal modes



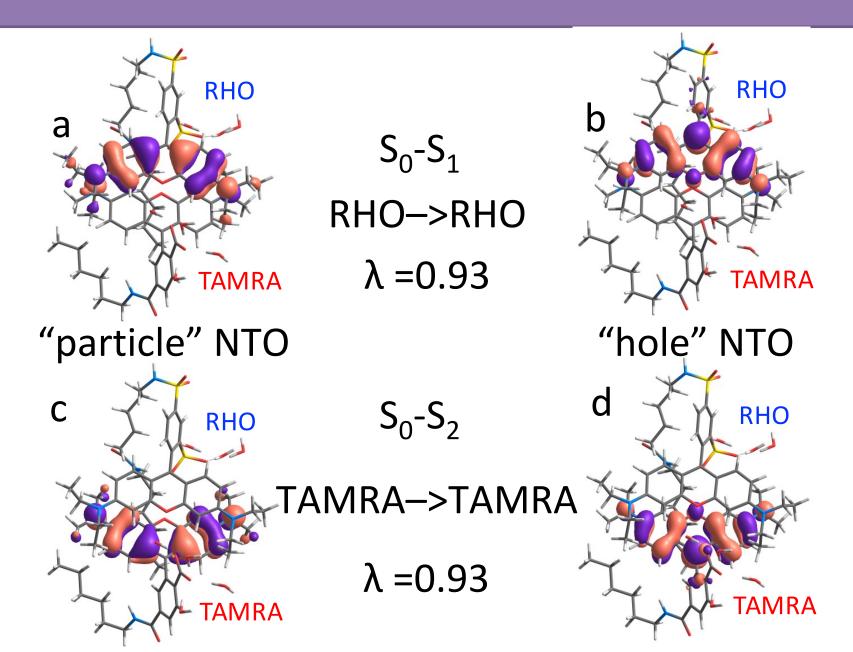
- Harmonic approximation is valid in the description of high-frequency optically active modes (S₂ absorption)
- There is a breakdown of the harmonic approximation for the S₁ absorption
- The Lineshape for the D1A dimer is narrower compare to the experimental absorption spectrum. Significant increase of the broadening of the S₀-S₁/S₀-S₂ bands in D1B dimer is mainly due to its more flexible structure: great number of soft modes with n<100cm⁻¹ contributes to its absorption profile.
- For these soft modes harmonic approximation can produce large errors

- Unbiased picking structures from the 8ns MD trajectory with the time step of 20ps (each 10th structure)
- For each sample only chromophore with linkers and nearest water molecules (within distance from the dye <2.5A) was taken

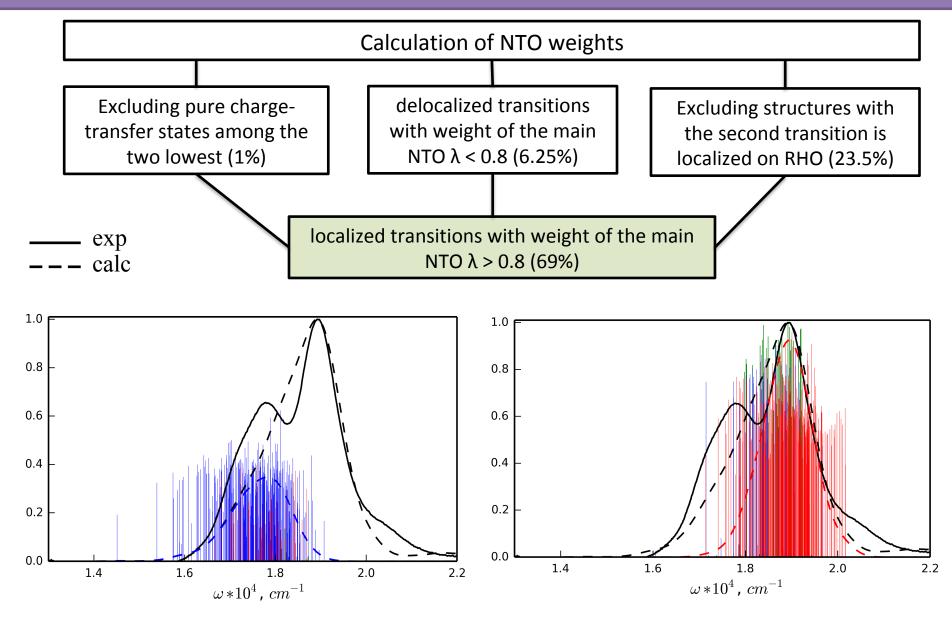


 TDDFT/ CAMD-B3LYP/6-311(d,p)/PCM for calculation of excitation energies and oscillator strengths

The dominant natural transition orbital pairs

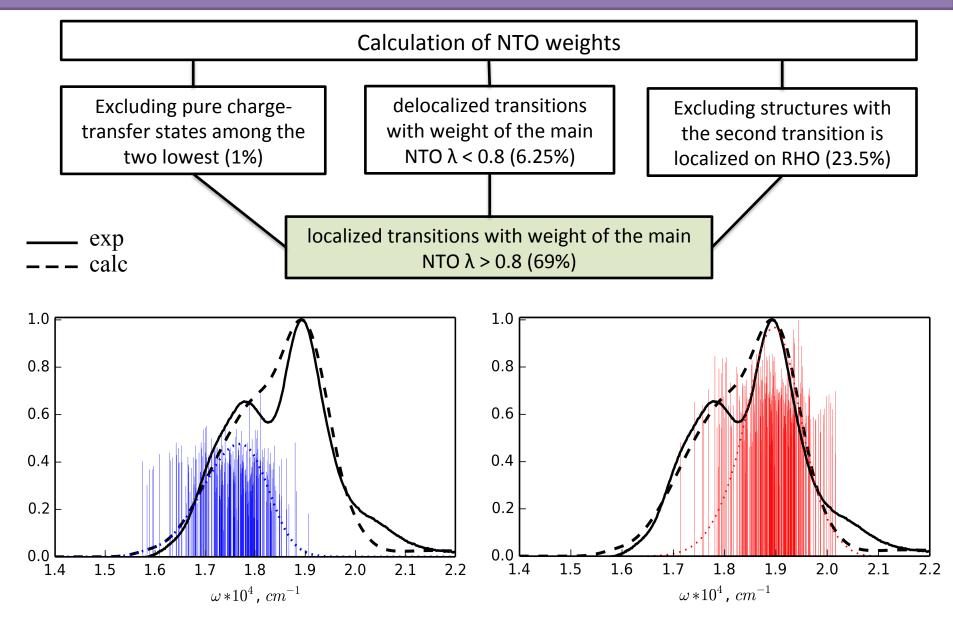


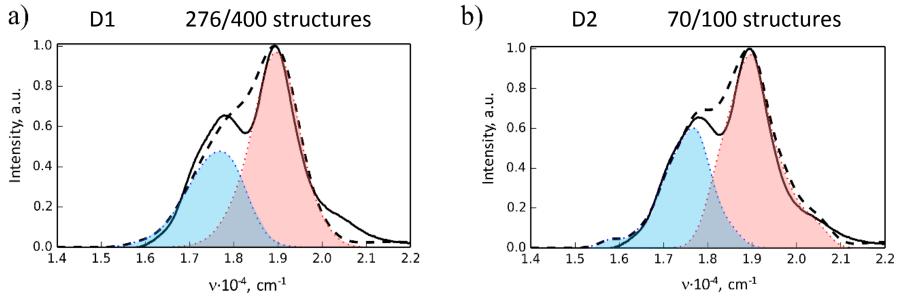
Sampling based on analysis of NTO weights (as a measure of the delocalization). 400 structures. D1



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Sampling based on analysis of NTO weights (as a measure of the delocalization). 400 structures. D1

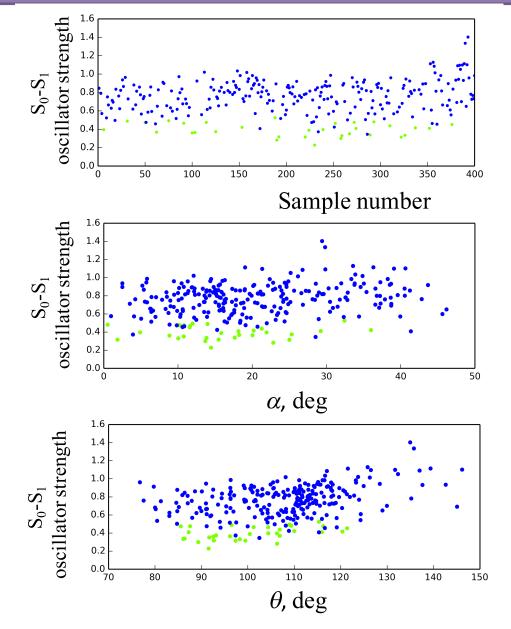




The color corresponds to the localization subunit: RHO (blue), TAMRA (red)

The calculated profiles are shifted to the positions of the experimental maxima (D = -613 cm^{-1} and D = -456 cm^{-1} for the D1 and D2 isomers respectively).

Correlation between the intermolecular angles and oscillator strength of the S0-S1 transition of D1 along MD trajectory.



(Top row) The oscillator strengths f of the S₀-S₁ transition for the conformations sampled from the D1 MD trajectory as calculated for the group with localized transitions only (276 samples). The sample number assigned sequentially for the 8ns MD trajectory with 20 ps step (400 samples in total).

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(Middle row) Correlation between the angle α between the polarization axes of the monomers and oscillator strength of the S₀-S₁ transition.

(Bottom row) Correlation of the angle θ between the polarization axis of RHO and line connecting the molecular centres in the RHO \rightarrow TAMRA direction and oscillator strength of the S₀-S₁ transition

The color of the dots denote the weight w of the NTO on RHO, green: w = 80-90%, blue dots: 90-100%.

Conclusions

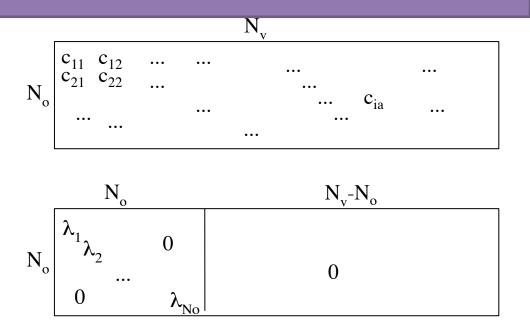
Conclusions

- Harmonic approximation is valid in the description of high-frequency hard optically active modes (S₂ absorption)
- For the **low-frequency** soft modes harmonic approximation can produce large errors
- The Lineshape for the D1A dimer is narrower compare to the experimental absorption spectrum. Significant increase of the broadening of the S₀-S₁/S₀-S₂ bands in D1B dimer is mainly due to its more flexible structure: great number of soft modes with n<100cm⁻¹ contributes to its absorption profile .
- Nuclear-ensemble sampling technique is developed. It is better in reproducing S₁ absorption band
- The origin of the high intensity of the first band is coming from the large electronic transition dipole for the dimers with realistic relative orientation between the monomers, when sterical effects of the complex media on the ground state geometry of the dimer are explicitly taken into account.

Natural Transition Orbitals (NTO)

- •Form transition density matrix **T**: the physically relevant quantity. $T_{ia} = \langle \Psi_{ex} | c_i^+ c_a | \Psi_0 \rangle$
- Diagonalize T T⁺ and T⁺T to obtain occupied and virtual NTOs

T T⁺ U_i =
$$\lambda_i$$
 U_i *i* = 1, n_{occ}
T⁺T V_i = λ_i V_i *i* = 1, n_{virt}

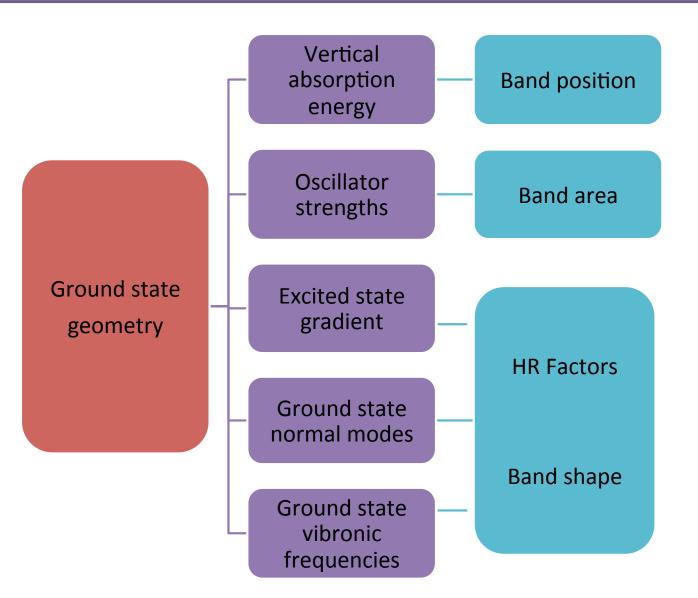


• Each occupied orbital is paired with single virtual orbital; the transition density is unchanged; the magnitude of λ shows how important it is to the transition.

"particle" NTO \rightarrow "hole" NTO $\lambda = \dots$

R.L. Martin, JCP 118, 4775 (2003). Batista and Martin, Encyclopedia of Computational Chemistry, 2004.

The flow chart of the lineshape calculations



Simulation of the absorption band based on Lax's model

$$\begin{split} I_{total}(\Omega) &= \sum_{i} w_{i} \sum_{b} f_{ab}^{i} I_{ab}^{i}(\Omega) \\ w_{i} &= \exp\left(-\Delta E_{i}/k_{B}T\right) / \sum_{j} \exp\left(-\Delta E_{j}/k_{B}T\right) \text{ the fraction of conformer (i)} \\ I^{i}(\Omega) &= \sum_{b} f_{ab}^{i} I_{ab}^{i}(\Omega) \quad f_{ab}^{i} \text{ oscillator strength} \\ I^{i}_{ab}(\Omega) &= \frac{1}{2\pi} \int_{-\infty}^{+\infty} \exp(-i\Omega t) G_{ab}(t) dt \quad \text{Intensity of individual transitions} \\ G_{ab}(t) &= \exp\left(i\Omega_{0}t - \sum_{j} S_{j}\left[(2\overline{n}_{j}+1)(1-\cos\Omega_{j}t)-i\sin\Omega_{j}t\right]\right) \\ \Omega_{0} \text{ 0-0 transition frequency } E_{0} &= \hbar\Omega_{0} \\ Q_{j} \text{ harmonic normal modes laying close to the initial conf.} \quad Q_{0} \\ Q_{j} \text{ vibrational frequencies} \\ S_{j} &= \Omega_{j}\Delta Q_{j}^{2}/2\hbar \text{ dimensionless Huang-Rhys (HR) parameters} \\ V_{j}^{a,b} &= \partial E^{a,b}/\partial Q_{j} \quad \text{gradients of potential surfaces along normal modes at initial configuration} \\ \Delta Q_{j} &= -(V_{j}^{b} - V_{j}^{a})/\Omega_{j}^{2} \quad \text{ origin shift in the normal coordinates} \\ \overline{n}_{j} &= \left[\exp(\hbar\Omega_{j}/k_{B}T) - 1\right]^{-1} \text{ the Planck's distribution} \\ \end{array}$$

Simulation of the absorption band based on Lax's model

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